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engineers, process engineers, and engineers for measurement and control, this resource explores the advanced simulation tools and techniques that are available to support experienced and aid new operators and engineers.

For first examination from 2022, these resources meet the real needs of the chemistry classroom. This practical write-in workbook is the perfect companion for the coursebook. It contains step-by-step guided investigations and practice questions for Cambridge International AS & A Level Chemistry teachers and students. Through practical investigation, it provides opportunities to develop skills- planning, identifying equipment, creating hypotheses, recording results, analysing data, and evaluating. The workbook is ideal for teachers who find running practical experiments difficult due to lack of time, resources or support. Sample data- if students can't do the experiments themselves - and answers to the questions are in the teacher's resource.

This title covers the entire syllabus for Cambridge International Examinations' International AS and A Level Biology (9700). It is divided into separate sections for AS and A Level making it ideal for students studying both the AS and the A Level and also those taking the AS examinations at the end of their first year. - Explains difficult concepts using language that is appropriate for students around the world - Provides practice throughout the course with carefully selected past paper questions at the end of each chapter We are working with Cambridge International Examinations to gain endorsement for this title.

This text presents a unified and up-to-date discussion of the role of atomic and molecular orbitals in chemistry, from the quantum mechanical foundations to the recent developments and applications. The discussion is mainly qualitative, largely based on symmetry arguments. It is felt that a sound mastering of the concepts and qualitative interpretations is needed, especially when students are becoming more and more familiar with numerical calculations based on atomic and molecular orbitals. The text is mathematically less demanding than most traditional quantum chemistry books but still retains clarity and rigour. The physical insight is maximized and abundant illustrations are used. The relationships between the more formal quantum mechanical formalisms and the traditional chemical descriptions of chemical bonding are critically established. This book is of primary interest to undergraduate chemistry students and others taking courses of which chemistry is a significant part.

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