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- No two electrons in an atom can have the same unique set of four quantum numbers.
- The first quantum number describes the principal energy level. For example, the quantum number 2 identifies the second principal energy level.

- The first two

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quantum numbers
together describe a
sublevel.

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theory. d.) it helps to
locate a proton in an
atom. d.) it is the same
as the Bohr's theory ...
the atomic nucleus b.)
atomic orbitals c.) atoms
d.) molecules.

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combustion. energy.

work. kinetic energy. the

process of burning

something that creates

energy. the capacity to

do work. what is done to

move an object against

some sort of

resistance.... the energy

of motion ($\frac{1}{2} mv^2$)

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Chapter 4. List the three statements that make up the modern atomic theory. 2. Define atomic number. What is the atomic number for a boron atom? 3. Which elements are alkali metals, alkaline earth metals, and halogens? 4. Two of the most well-

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known isotopes of carbon are carbon-12 and carbon-14.

4.E: Homework

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the electron are also
called orbitals. The
orbital shown in Figure
11.3 is called the 1s
orbital. ~ e negative
charge is most intense at

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the nucleus and
decreases in intensity
with distance outward.

Chapter 11 Modern
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Chapter 4 Atomic
Structure Section 4.3

Modern Atomic Theory
(pages 113-118). Use this
table to answer
questions 12 and 13..

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Worksheet. Answer the following questions, circle the best answer.

Chapter 4 Atomic Structure Worksheet Answer Key Pdf

The modern atomic theory states that atoms of one element are the same, while atoms of different elements are different. What makes atoms of different

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elements different? The fundamental characteristic that all atoms of the same element share is the number of protons. All atoms of hydrogen have one and only one proton in the nucleus; all atoms of iron have 26 protons in the nucleus.

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separate atoms (Figure
4.1). Radiant energy
from the sun causes
sunburn 4.1 Energy

Some reactions absorb
energy. Energy from the
sun $O_2(g) + \text{energy}$
 $2O(g)$ Figure 4.1 Some
reactions absorb energy.

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120 Chapter 4 Modern Atomic Theory Before we can begin to explain the role that energy plays in these and other chemical

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