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a) O atom = 0, C atom = 0, and N atom =
0 b) O atom = 0, C atom = 0, and N atom
= - c) O atom = -1, C atom = 0, and N
atom = 0 d) O atom = -1, C atom = -1,
and N atom = + e) O atom = +1, C atom =
0, and N atom = -. 17. Use VSEPR to
predict the electron-pair geometry and the
molecular geometry of sulfur dioxide, SO
2.

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that portion of chemistry dealing with numerical relationships in chemical reactions; the calculation of quantities of substances involved in chemical equations

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 10^{-34} J s (Planck's Constant) $c = 3.00 \times$
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10-7 m-1 (Rydberg Constant) Multiple Choice (5 points each) Identify the letter of the choice that best completes the statement or answers the question and legibly write the letter

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