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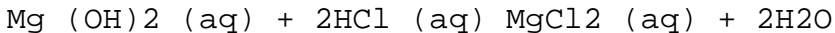
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NIE:  $2\text{OH}^{-} (\text{aq}) + 2\text{H}^{+} (\text{aq}) \rightarrow 2\text{H}_2\text{O} (\text{l})$   
(your final answer would be:  $\text{OH}^{-} (\text{aq}) + \text{H}^{+} (\text{aq}) \rightarrow \text{H}_2\text{O} (\text{l})$ ) 4.

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