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*Conservation of
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*(Reactants =
Products)* **How to**

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Answers Author:
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Balancing

equations Get

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Erase the

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formulas in each
text box.

Introduction: In
a chemical
reaction, the
reactants are
the substances
that enter into
the reaction,
and the products
are the
substances that
are made in the

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Answers ...

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Determining
concentration
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ready: Click
Reset. Select
1.00 M NaOH for
the Burette and
Mystery H₂SO₄

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Equations for the Flask.

Select

Bromthymol blue

for the

Indicator.

Introduction:

Adding a drop of

strong acid or

base into a

neutralized

solution is

similar to

adding a drop of

strong acid or

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base to water –
it causes an
abrupt change in
pH.

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equation in the

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Equations Gizmo,

type the

chemical

formulas into

the text boxes

of the Gizmo.

First, type in

"H₂+O₂" in the

Reactants box

and "H₂O" in the

Products box.

This...

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Chemical
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(ANSWER KEY) by**

...

Using the
chemical
equations in the
Gizmo as a
guide, match the
following
definitions to
the type of

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Equations. b _____
one reactant is
Activity B
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broken down into
two or more

products. _a_____

Two or more
reactants
combine to form
one product.

d two
compounds react
to form two
different
compounds.

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To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in "H₂+O₂" in the

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Reactants box
and "H₂O" in the
Products box.
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circumstances

surrounding

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cause. I began

to fight for me.

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an Integrated

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The Equation Is

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Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical

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Equations into
the text boxes
of the Gizmo.
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First, type in
“H₂+O₂” in the
Reactants box
and “H₂O” in the
Products box.
This represents
the reaction of
hydrogen and
oxygen gas to
form water.

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Answers To Activity C

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Methane reaction

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ready: Select C

+ H₂ becomes

CH₄. Goal:

Predict the

results of a

chemical

reaction. 1.

Count: CH₄ is

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the chemical formula for methane. In order to produce a single molecule of methane, how many carbon (C) atoms are needed? _____
Hydrogen atoms? _____ 2.

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