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States of Matter. Terms in this set (32) kinetic energy. The energy an object has because of its motion. kinetic theory. The tiny particles in all
forms of matter are in constant motion.

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Chapter 13 – Gases 195 Exercise 13.3 – Equation Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which
is about 4.3% carbon. $2C + O_2 \rightarrow 2CO$ $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ $2CO + C \rightarrow 3C$ (in iron) CO_2 Pig iron is easier to shape than pure iron, and the presence
of carbon lowers its melting point

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changes that lead to greater dispersal of matter are likely to take place. To explain why liquids tend to mix and form solutions. To show how
you can predict whether substances are soluble in water.

Chapter 13 Solution Dynamics - An Introduction to Chemistry

Name Date Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery Section 13.3 Liquids and Solids In your
textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false, change the italicized
word or phrase to make it true.

13 STUDY GUIDE FOR CONTENT MASTERY 13 STUDY GUIDE FOR ...

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the sides question what are the instruments that are used to

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CHAPTER 13 Table Of Contents Click a hyperlink to view the corresponding slides. •State the relationships among pressure, temperature,
and volume of a constant amount of gas.

Chemistry: Matter and Change

196 Study Guide for An Introduction to Chemistry Chapter 13 Map Chapter Checklist Read the Review Skills section. If there is any skill
mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the
lecture that describes it.

Chapter 13 - Gases - An Introduction to Chemistry

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