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Properties Matter – Anything that has
MASS and VOLUME. ATOMS and

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MOLECULES the basic building blocks of matter. **ATOM**-- the smallest unit of an element that still maintains the properties that element. **ELEMENT**– a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1

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Properties of Matter pages 70–75 Problem-
Solving Lab 1. Explain why the flow of a
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practical and safe use.

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5 25 Electrons in Atoms Reviewing

Vocabulary Match the definition in

Column A with the term in Column B.

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Change Chapter B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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extensive property. intensive property.

extensive properties (examples) intensive

property (examples) a property that

depends on the amount of matter in a

sample. a property that depends on the

type of matter in a sample not.... mass,

volume, length, weight.

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11 31 number of moles of known

substance C H C H C H O O O CO CO

CO H O H O H O number of moles of

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unknown substance O CO H O CH CO H
 $\text{O CH O H O C H O CO}$ use mole ratio 5
mol O_2 1 mol C_3H_8 3 mol CO_2 1 mol
 C_3H_8 4 mol H_2O 1 mol C_3H_8 1 mol
 C_3H_8 5 mol O_2

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A “chemical change” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different

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