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NaCl(aq) + AgNO3(aq) - AgCl(s) + NaNO3(aq) H2SO4(aq) + 2NaOH(aq) - N2S04(aq) + 2H2O(l) N2S03(aq) + 2HCl(aq) - 2NaCl(aq) + Hp(l) + SO2(g) A compound burns in the presence of oxygen, producing energy in the form of heat and light. The combustion of organic compounds produces carbon dioxide and water.

Classifying Chemical Reactions—60Webs

Summary. In this inquiry-based experiment, the students design and carry out a sequence of chemical reactions for the separation and identification of iron, silver, and zinc ions in water. Two parallel series of tests will be carried out, one using a known sample containing all three metal cations and the other using unknown samples containing one, two, or all three cations.