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this as you study

Chemistry II. It has

the form: In this

equation  $\Delta G^\circ$  is the

Gibbs Free Energy of

a reaction under

standard conditions of

1 atm (or 1 bar) for

gases and 1 M for

solutions at 25°C;

$\Delta H^\circ$  is the enthalpy of

the reaction under

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standard conditions;  $T$  is the Kelvin temperature; and  $\Delta S^\circ$  is the entropy of the reaction under standard conditions.

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ion. A species of an  
element in which the  
number of electrons  
does not equal the  
number ...

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