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a) How would you record the following volume measurement on a burette shown in the picture using rules of significant figures?

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07. mL $V = b$) If the burette was used to dispense a certain amount of solution, and the new volume reading is shown in the picture below, what volume of the solution was dispensed?

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Answers. Carbon atoms bond reasonably strongly with other carbon atoms. Carbon atoms bond reasonably strongly with atoms of other elements.

Carbon atoms make a

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large number of
covalent bonds (four).

2. Organic
compounds are based
on covalent bonding
or electron sharing.

The atoms C, H, O, N
that make up organic
compounds are all
nonmetals. 3.

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Hydrocarbons are molecules that contain only hydrogen and carbon atoms. Each Carbon atom forms 4 bonds and each hydrogen forms 1 bond. Hydrocarbons include (among other things) alkanes - all single bonds between

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carbons alkenes - one or more double bonds between carbons
alkynes - one or more triple bonds between carbons
aromatics - contain benzene ring or similar structures.

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