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Chapter 9 Reading Question 1 Part A
What is a possible mole-mole factor for the following equation?
 $5 \text{ C} + 2 \text{ SO}_2 \rightarrow \text{CS}_2 + 4 \text{ CO}$
Hint 1. Mole-mole factors come from the coefficients of a balanced chemical equation. See Section 9.2 (Page 252).

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Chapter 9

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Chapter 9 Problem 9.113 Problem

9.113 Assume that you have 1.95 g of nitroglycerin in a 605.5 mL steel container at 20.0 °C and 1.00 atm pressure. An explosion occurs, raising the temperature of the container and

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the central science; its mastery
enables an easy path to physics,
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Answers. To understand chemistry is to understand its principles and then apply them

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mole factors come from the

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Answers of a balanced chemical equation. See Section 9.2 (Page 252) .

ANSWER: Correct Chapter 9 Reading Question 2 Part A Iron reacts with oxygen in the air to produce rust (Fe_2O_3)
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Answers

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Chapter 3 Mastering chemistry answers chapter 3. Learn vocabulary, terms and more with flashcards, games and other study tools. On the periodic table, one of the columns within the main group elements; a family or group of elements exhibits similar chemical properties. (3

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