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B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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A, continued Use this figure to answer questions 7 and 8. \_\_\_\_\_ 7.  
A solution containing 35 g of  $\text{Li}_2\text{SO}_4$  dissolved in 100 g of water is heated from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ . According to information in the figure, this temperature change would result in a. an additional 5 g of  $\text{Li}_2\text{SO}_4$  in ...

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18. Dalton's atomic theory acyreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have

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the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron  
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