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Organic chemistry - laboratory methods 3 Grading and Rules of evaluation The final grade will be weighted arithmetic mean of two grading elements: (1) lab work (60%) and (2) post-lab reports (40%) lab work will be graded twice - after 4th and 8th class; the lab work grade will be an arithmetic mean of those two grades

Organic chemistry Laboratory methods

In addition to General, Organic and Biological Chemistry, 6th Edition, she is also the author of An Introduction to General, Organic, and Biological Chemistry, 13th Edition, with the accompanying Study Guide and Selected Solutions Manual, Laboratory Manual and Essentials Laboratory Manual, and Basic Chemistry, 5th Edition, with the accompanying ...

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Preface To the Instructor Acknowledgments Introduction Problem Solving in the Organic Chemistry Laboratory Scientific Methodology Organization of This Book A Guide to Success in the Organic Chemistry Lab Laboratory Safety Safety Standards Protecting Yourself Preventing Laboratory Accidents Reacting to Accidents: First Aid Reacting to Accidents: Fire Chemical Hazards Finding and Using Chemical Safety Information Chemistry and the Environment Disposal of Hazardous Wastes Green Chemistry Part I Mastering the Operations 1 The Effect of pH on a Food Preservative 2 Separating the Components of "Panacetin"; 3 Identifying a Constituent of "Panacetin"; 4 Synthesis of Salicylic Acid from Wintergreen Oil 5 Preparation of Synthetic Banana Oil 6 Separation of Petroleum Hydrocarbons 7 A Green Synthesis of Camphor 8 Identification of a Petroleum Hydrocarbon 9 Isolation and Isomerization of Lycopene from Tomato Paste 10 Isolation and Identification of the Major Constituent of Clove Oil 11 Identification of Unknown Ketones 12 The Optical Activity of α -Pinene: A Chemical Mystery Part II Correlated Laboratory Experiments 13 Investigation of a Chemical Bond by Infrared Spectrometry 14 Properties of Common Functional Groups 15 Thin-Layer Chromatographic Analysis of Drug Components 16 Separation of an Alkane Clathrate 17 Isomers and Isomerization Reactions 18 Structures and Properties of Stereoisomers 19 Bridgehead Reactivity in an S_N1 Solvolysis Reaction 20 Reaction of Iodoethane with Sodium Saccharin, an Ambident Nucleophile 21 Dehydration of Methylcyclohexanols and the Evelyn Effect 22 Testing Markovnikov's Rule 23 Stereochemistry of Bromine Addition to trans-Cinnamic Acid 24 A Green Synthesis of Adipic Acid 25 Preparation of Bromotriphenylmethane and the Trityl Free Radical 26 Chain-Growth Polymerization of Styrene and Methyl Methacrylate 27 Synthesis of Ethanol by Fermentation 28 Reaction of Butanols with Hydrobromic Acid 29 Borohydride Reduction of Vanillin to Vanillyl Alcohol 30 Synthesis of Triphenylmethanol and the Trityl Carbocation 31 An Unexpected Reaction of 2,3-Dimethyl-2,3-butanediol 32 Identification.

This comprehensive laboratory text provides a thorough introduction to all of the significant operations used in the organic lab and includes a large selection of traditional-scale and microscale experiments and minilabs. Its unique problem-solving approach encourages students to think in the laboratory by solving a scientific problem in the process of carrying out each experiment. The Second Edition contains a new introductory section, "Chemistry and the Environment," which includes a discussion of the principles of green chemistry. Several green experiments have been added, and some experiments from the previous editions have been revised to make them greener.

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This advanced textbook for teaching and continuing studies provides an in-depth coverage of modern food chemistry. Food constituents, their chemical structures, functional properties and their interactions are given broad coverage as they form the basis for understanding food production, processing, storage, handling, analysis, and the underlying chemical and physical processes. Special emphasis is also given to food additives, food contaminants and the understanding the important processing parameters in food production. Logically organized (according to food constituents and commodities) and extensively illustrated with more than 450 tables and 340 figures this completely revised and updated edition provides students and researchers in food science or agricultural chemistry with an outstanding textbook. In addition it will serve as reference text for advanced students in food technology and a valuable on-the-job reference for chemists, engineers, biochemists, nutritionists, and analytical chemists in food industry and in research as well as in food control and other service labs.

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Standard medicinal chemistry courses and texts are organized by classes of drugs with an emphasis on descriptions of their biological and pharmacological effects. This book represents a new approach based on physical organic chemical principles and reaction mechanisms that allow the reader to extrapolate to many related classes of drug molecules. The Second Edition reflects the significant changes in the drug industry over the past decade, and includes chapter problems and other elements that make the book more useful for course instruction. New edition includes new chapter problems and exercises to help students learn, plus extensive references and illustrations. Clearly presents an organic chemist's perspective of how drugs are designed and function, incorporating the extensive changes in the drug industry over the past ten years. Well-respected author has published over 200 articles, earned 21 patents, and invented a drug that is under consideration for commercialization

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Ideal for those who have previously studied organic chemistry but not in great depth and with little exposure to organic chemistry in a formal sense. This text aims to bridge the gap between introductory-level instruction and more advanced graduate-level texts, reviewing the basics as well as presenting the more advanced ideas that are currently of importance in organic chemistry. * Provides students with the organic chemistry background required to succeed in advanced courses. * Practice problems included at the end of each chapter.

100 sheets of carbonless graph paper, lay-flat plastic-coil binding, laboratory safety document, reference tables.

Learn the secrets of soil chemistry and its role in agriculture and the environment. Examine the fundamental laws of soil chemistry, how they affect dissolution, cation and anion exchange, and other reactions. Explore how water can form water-bridges and hydrogen bonding, the most common forces in adsorption, chelation, and more. Discover how electrical charges develop in soils creating electrochemical potentials forcing ions to move into the plant body through barriers such as root membranes, nourishing crops and plants. You can do all this and more with Principles of Soil Chemistry, Fourth Edition. Since the first edition published in 1982, this resource has made a name for itself as a textbook for upper level undergraduates and as a handy reference for professionals and scientists. This fourth edition reexamines the entire reach of soil chemistry while maintaining the clear, concise style that made previous editions so user-friendly. By completely revising, updating, and incorporating a decade's worth of new information, author Kim Tan has made this edition an entirely new and better book. See what's new in the Fourth Edition Reexamines atoms as the smallest particle that will enter into chemical reactions by probing new advances testifying the presence of subatomic particles and concepts such as string theory Underscores oxygen as the key element in soil air and atmosphere for life on earth Reevaluates the idea of transformation of orthoclase into albite by simple cation exchange reactions as misleading and bending scientific concepts of ion exchange over the limit of truth Examines the role of fertilizers, sulfur, pyrite, acid rain, and nitrogen fixation in soil acidity, underscoring the controversial effect of nitrification on increasing soil acidity over time Addresses the old and new approaches to humic acids by comparing the traditional operational concept against the currently proposed supramolecular and pseudomicellar concept Proposes soil organics, such as nucleic acids of DNA and others, to also adsorb cation ions held as diffusive ion clouds around the polymers Tan explains, in easy and simple language, the chemical make-up of the four soil constituents, their chemical reactions and interactions in soils as governed by basic chemical laws, and their importance in agriculture,

industry, and the environment. He differentiates soil chemistry from geochemistry and physical chemistry. Containing more than 200 equations, 123 figures, and 38 tables, this popular text and resource supplies a comprehensive treatment of soil chemistry that builds a foundation for work in environmental pollution, organic and inorganic soil contamination, and potential ecological health and environmental health risks.

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