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Chemical Reactions
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Chemical reaction can be concisely represented by a chemical equation. The substances that undergo a chemical change are the reactants. The new substances formed in a chemical reaction are the products.

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Chemistry. chemical equation. skeleton
equation. catalyst. coefficients. a
representation of the chemical reaction; the
formulas of the reactants and products.
... a chemical equation that
does not indicate the relative amount of
reactants and products.
... a
substance that speeds up a reaction but is
not used up in the reaction.

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a chemical change in which one element replaces a second element in a compound;
displacement reaction Activity Series a list of elements in order of decreasing activity

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represented by a chemical 1. The substances

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that undergo a chemical change are the 2.
The new substances formed in a chemical
reaction are the 3.

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- A skeleton equation is a chemical equation that does not indicate the relative amounts of the reactants and products. Like a recipe with no amounts listed.

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Answers

11.2 Types of Chemical Reactions > 13 A decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or more products. • The products can be any combination of elements and compounds. • Most decomposition reactions require energy

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