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1s²2s²2p⁶3s¹ Mg

1s²2s²2p⁶3s² Al

1s²2s²2p⁶3s²3p¹ Si

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Is_{22s22p63s23p2} P

1s_{22s22p63s23p3} 26 24

S 1s_{22s 2P 3s 3p} CI

1s_{22s22p63s23p5} Ar

1s_{22s22p63s23p6} 1 1 4

2 59. 4 62. 2 65. 2 68. 2

The mixture contains at least three parts. One solid remained on the filter paper. The other two

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same group of the
period table behave
similarly because they
have the same number
of what? (octet rule,
valence electrons,
electron dot structures,
cations, group) Valence
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