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3. Apply How many atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits. 1.007

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g 1 H-1 atom $\frac{1.672 \times 10^{-24} \text{ g}}{1.672 \times 10^{-24} \text{ g a.}}$
6.023 $\times 10^{23}$ H-1 atoms 4.

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30. Compare and contrast sublimation

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and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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Chapter 9 141 Section 9.1 Reactions
and Equations pages 282–288

Practice Problems pages 284–287

Write skeleton equations for the following word equations. 1. Hydrogen and bromine gases react to yield hydrogen bromide.

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