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## Answers Chemistry Answers

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### WORKSHEET Ionic Bonds Use with

Chapter 7, Section 7.2 1. How many valence electrons does a neutral magnesium (Mg) atom have? 2. What is the charge on a magnesium ion?

What does magnesium have to do to

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form such an ion, and why does it  
tend to do so? 3. How many valence  
electrons does a single neutral  
chlorine (Cl atom) have? 4.

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Chemistry: Matter and Change

Chapter 5. Quest 11 Name: Date: Hour:

Information: Energy Levels and

Sublevels ... Considering your answer

to question 5, how many electrons

can fit in the entire 5<sup>th</sup> energy level?

can (Ave is a ha AC5a) 5<sup>th</sup> 7. Write

the notation for an electron spinning

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clockwise in a p-sublevel...

## Chapter 9

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WORKSHEET The Activity Series Use  
with Chapter 9, Section 9.2 1. For each  
of the following pairs of elements,

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underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

TEACHING TRANSPARENCY

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WORKSHEET The Activity Series

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Chemistry: Matter and Change 0

Chapter 8 . 1.7, the CF bond is polar

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covalent. (b)  $EN_{Cl} - EN_{P} = 3.0 - 2.1 = 0.9$ . Since  $0.5 < EN_{Cl} - EN_{P} < 1.7$ , the PCl bond is polar covalent. (c)  $EN_{Cl} - EN_{Li} = 3.0 - 1.0 = 2.0$ . Since  $EN_{Cl} - EN_{Li} > 1.7$ , the bond is ionic. Section 8.5 Electronegativity and Polarity 265 Compare and contrast polar covalent and ionic bonds. that is in excess.

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11 - Teacher Worksheets

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is the shape of an s orbital? Class 16  
Use with Chapter 5, Section 5.2 2.

What is the relationship between the  
size of an s orbital and the principal

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energy level in which it is found? 3.  
What is the shape of a p orbital? How  
many p orbitals are there in a ulevel?  
4.

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