

## Technical Chemistry Gas Laws Magic Square Answers

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Technical Chemistry: Gas Laws Name: Match the variables used to describe gases to the correct unit. 1. 2. 4. 5 kPa rnL K mm Hg  
atmospheres (atm) L a. pressure b. temperature c. volume Complete the following statements by writing "decreases," "increases," or  
"remains the same" on the line provided. As a gas is compressed in a cylinder 9. its mass

Region 14 - Bethlehem & Woodbury Connecticut

Technical Chemistry - Gas Laws Magic Square You must show your work in the square. Name A. A sample of neon gas occupies a volume of  
2.8 L at 1.8 atm. What would its volume be at 1.2 atm? B. A balloon full of air has a volume of 2.75 L at a temperature of 18 ° C. What Ois  
the balloon's volume at 45 C? C. If 3.0 L of a gas at heated to 30.0 ° C

O 3L - Ms Galloway

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Name&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr;&mlr; A. A sample of neon gas occupies a volume of 2.8 L at 1.8 atm. What would its volume be at 1.2 atm? B. A balloon full of air has a volume of 2.75 L at a temperature of 18 o C.

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The three fundamental gas laws discover the relationship of pressure, temperature, volume and amount of gas. Boyle's Law tells us that the volume of gas increases as the pressure decreases. Charles' Law tells us that the volume of gas increases as the temperature increases. And Avogadro's Law tell us that the volume of gas increases as the amount of gas increases. The ideal gas law is the combination of the three simple gas laws.

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